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3. Strong Acid 4. Weak Base 5. Weak Acid A.
The measure of acids

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and bases. B. A compound that adds a few OH^- ions to water.

C. The product of a neutralization reaction between an acid and a

base. D. A compound that adds a few H^+

ions to water. E. A compound that adds a lot of H^+ ions to water.

pH and Acid Rain

Plants and animals

need water

**Acids and Bases - Mr
Murray's Science**

Page 5/26

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Acids And Bases
and Music

Acids and bases Acids are chemicals that add H^+ ions to water when you make a solution. Bases are chemicals that add OH^- ions to water in a solution. In other words when you put a chemical into water and stir it up, if it makes H^+ ions, it is an acid; if it makes OH^- ions: it is a base. If an acid makes a lot of H^+ ions, we call it a strong acid, like HCl .

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When acids and bases
are mixed they
neutralize each other.

... C. Stephen Murray

Unit 5: 3 1. Acid 2.

Base 3. Neutral 4.

Neutralize 5. Acid Rain

6. A Salt 7. Table salt

A. To mix acids and
bases to cancel each
other out and make
salt water. B. A

compound that adds

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H⁺ ions to water.
Answer Key

Strong acids

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Naming Compounds

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m

Goal 2.2.6 • Recognize common inorganic acids including hydrochloric (muriatic) acid, sulfuric acid, acetic acid, nitric acid and citric acid. •

Recognize common bases including sodium bicarbonate, and hydroxides of sodium, potassium, calcium, magnesium, barium and ammonium. •

Define acids and bases according to the

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Arrhenius theory. •
Develop an
understanding of the
pH scale and the ...

**Unit 2.2.6 Acids and
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