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E1-22 First find the "logarithmic average" by $\log_d a_v = \frac{1}{2} \log(2 \cdot 1026) + \log(1 \cdot 10^{15})$. $\frac{1}{2} \log 2 \cdot 1026 + \frac{1}{2} \log 10^{15} = \log 2 \cdot 10^{11} = \log p \cdot 2 \cdot 10^{11}$. Solve, and $d a_v = 450 \text{ km}$. E1-23 The number of atoms is given by $(1 \text{ kg}) / (1.00783 \cdot 1.661 \cdot 10^{-27} \text{ kg})$, or $5.974 \cdot 10^{26}$ atoms.

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David Halliday is associated with the University of Pittsburgh as Professor Emeritus. As department chair in 1960, he and Robert Resnick collaborated on Physics for Students of Science and Engineering and then on Fundamentals of Physics. Fundamentals is currently in its eighth edition and has since been handed over

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