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AND THE PERIODIC
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Given the following isotopes, determine the atomic number, the mass number, the

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number of protons,
electrons and
neutrons. Isotope
Symbol ^{131}I ^{31}S

|He SAI

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Science / Chapter 4 - atomic structure

Section 4.2 Structure of
the Nuclear Atom One
change to Dalton's
atomic theory is that
atoms are divisible into
subatomic particles:
Electrons, protons, and

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neutrons are examples of these fundamental particles. There are many other types of particles, but we will study these three.

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TABLE CHAPTER 4
WORKSHEET PART A
Given the following isotopes, determine

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the atomic number, the
mass number, the
number of protons,
electrons and

neutrons. Isotope

Symbol ^{131}I ^{53}I ^{238}U

^2He ^{81}W ^{36}Kr ^{14}C

37K Atomic Number

Mass Number Protons

Electrons Neutrons

Isotope Name PART B.

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Structure Worksheet .
Answer the following
questions, circle the
best answer. 1) Which
of the following are
isotopes of each other?

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- a) ^{14}C and ^{14}N b) ^3H
and ^4He c) ^2H and ^1H
d) none of these . 2)

The net charge on an atom that has 13 protons, 12 neutrons, and 10 electrons is

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Structure Answers ATOMIC STRUCTURE AND THE PERIODIC TABLE CHAPTER 4 WORKSHEET PART A

Given the following isotopes, determine the atomic number, the mass number, the number of protons, electrons and neutrons

Isotope	Symbol	Atomic Number	Mass Number	Protons	Electrons	Neutrons
131I						
53I						
2He						
81W						
36Kr						
37K						

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TABLE CHAPTER 4
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Given the following

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isotopes, determine
the atomic number, the
mass number, the
number of protons,
electrons and neutrons
Isotope Symbol ^{131}I
 ^{53}I ^{81}W
 ^{36}Kr ^{37}K Atomic
Number Mass Number
Protons Electrons
Neutrons Isotope

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Structure Worksheet .

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Answer the following questions, circle the best answer. 1) Which of the following are isotopes of each other?
a) ^{14}C and ^{14}N b) ^3H and ^4He c) ^2H and ^1H d) none of these . 2)

The net charge on an atom that has 13 protons, 12 neutrons, and 10 electrons is . a) $5+$ b) -3 c) $3+$ d) -9

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This unit's focus is on
atomic structure.
Students will cover
concepts related to
atomic mass, isotopes,

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electron configurations
and periodicity.

Chapter 4 - Atomic
Structure

Unit 2 - Chapters 4, 5 & 6 - Mrs. Gingras' Chemistry Page

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