

Chapter 15 Test Acids And Bases Answers

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Chapter 15 Test Acids And Bases Answers
Chapter 15 Test Acids And Bases Answers when acids or bases act as acids or bases, they form their conjugate and the strength is the inverse of their parent. and there are certain strong acids and strong bases. LiOH, NaOH, KOH - when act as bases (get rid of H+ ion) then left with Li Page 13/24

Chapter 15 Test Acids Bases Answers
Chapter 15: Acids and Bases Acids and Bases Arrhenius Definitions: acids - compounds that produce an increase in [H+] when dissolved in water bases - compounds that produce an increase in [OH-] when dissolved in water Lewis Definitions: acids - electron pair acceptors bases - electron pair donors Bransted-Lowry Definitions: acids - H+ donors

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Chapter 15 Test Acids And Bases Answers
Chapter 15 Acids, Bases and Salts. 2 15.1 Acids and bases acid derived from Latin acidus (meaning sour or tart) related to Latin acetum (meaning vinegar) characteristic properties associated with acid: 1. sour taste 2. change the color of litmus from blue to red 3. react with .metal (such as Zn.

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Chapter 15 Test Acids And Bases Answers
Chapter 15 Test Acids And 454 CHAPTER 15 Formula Acid name HF hydrofluoric acid HCl hydrochloric acid HBr hydrobromic acid HI hydriodic acid H 2S hydrosulfuric acid TABLE 15-1 Names of Binary Acids FIGURE 15-2 A strip of pH paper dipped into vinegar turns red, show-ing that vinegar is an acid. Binary Acid Nomenclature 1.

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